

Al-Sc-Zn (Aluminum-Scandium-Zinc)

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The phase equilibria of this system were studied by [1990Ikr], [1995Gan] and [2004Rok].

Binary Systems

The Al-Sc phase diagram [1999Cac, Massalski2] depicts the following intermediate compounds: ScAl_3 ($L1_2$, AuCu_3 -type cubic), ScAl_2 ($C15$, MgCu_2 -type cubic), ScAl ($B2$, CsCl -type cubic), and Sc_2Al ($B8_2$, Ni_2In -type hexagonal). In the Al-Zn system [Massalski2], solidification occurs through a eutectic reaction at 381 °C yielding (Zn) and (Al).

On solidification, (Al) contains 67 at.% of dissolved Zn. At lower temperatures, this solid solution exhibits a miscibility gap, with a monotectoid reaction at 316 °C: $(\text{Al})' \leftrightarrow (\text{Al}) + (\text{Zn})$. The Sc-Zn system was investigated in the 40-100 at.% Zn range by [1997Pal]. The intermediate phases found are: ScZn_{12} ($D2_b$, ThMn_{12} -type tetragonal), $\text{Sc}_3\text{Zn}_{17}$ ($\text{Ru}_3\text{Be}_{17}$ -type cubic), $\text{Sc}_{13}\text{Zn}_{58}$ ($\text{Gd}_{13}\text{Cd}_{58}$ -type hexagonal), ScZn_2 ($C32$, AlB_2 -type hexagonal), and ScZn ($B2$, CsCl -type cubic).

Ternary Phase Equilibria

The solidification characteristics in the Sc-lean region of this system were studied by [1995Gan]. Microstructural examination and differential thermal analysis were employed. The vertical section determined by [1995Gan] along the Zn- ScAl_2 join is shown in Fig. 1. This pseudobinary section is of the simple eutectic type, with the eutectic reaction e_3 at 390 °C and at the composition (at.%) 6.7Al-3.3Sc-90Zn. A liquidus projection in the Sc-lean region was also determined by [1995Gan]. This is shown in Fig. 2. Two four-phase invariant reactions occur in this region. The transition reaction U: $L + \text{ScAl}_2 \leftrightarrow \text{ScAl}_3 + (\text{Zn})$ occurs at 370 °C and at 9.5Al-2.0Sc-88.5Zn (at.%). The final solidification is through the ternary eutectic reaction E: $L \leftrightarrow (\text{Al}) + (\text{Zn}) + \text{ScAl}_3$ at 367 °C and at 10.0Al-1.7Sc-88.3Zn (at.%).

According to [1995Gan], two isothermal sections at 500 and 300 °C were determined for this system by [1990Ikr]. The reference [1990Ikr] is not available to this reviewer. Recently, [2004Rok] determined an isothermal section for this system at 500 °C in the Al-rich region. With starting metals of 99.99% Al, 99.86% Sc, and 99.99% Zn, [2004Rok] melted alloys in a resistance furnace under a flux cover. The alloys were annealed at 500 °C for 50 h and quenched in water. The phase equilibria were studied with

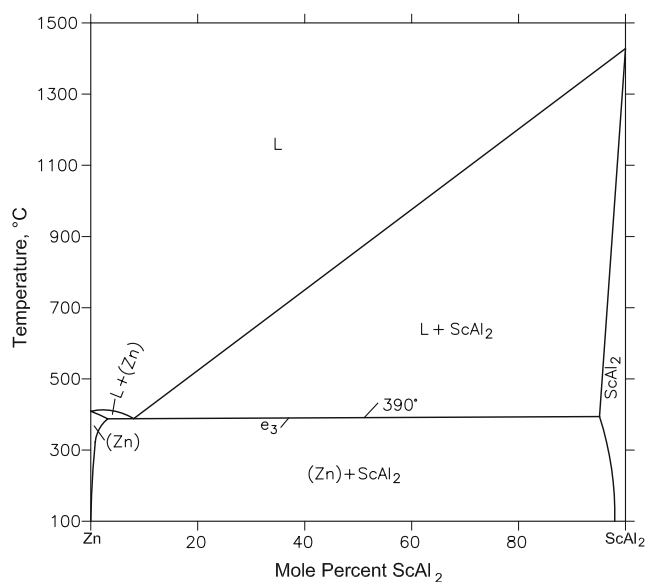


Fig. 1 Al-Sc-Zn pseudobinary section along the Zn- ScAl_2 join [1995Gan]

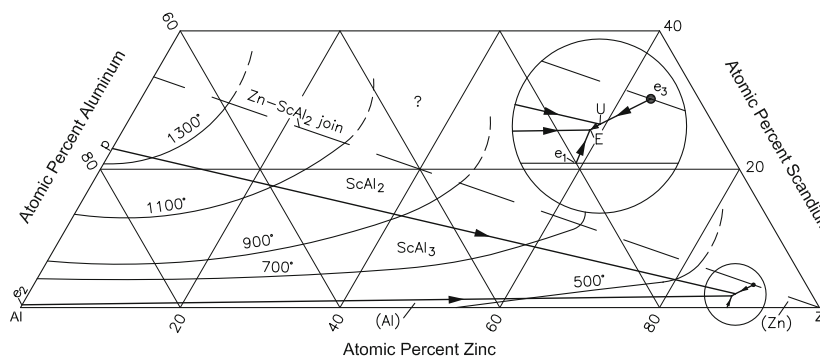


Fig. 2 Al-Sc-Zn liquidus projection in the Sc-lean region [1995Gan]

Section II: Phase Diagram Evaluations

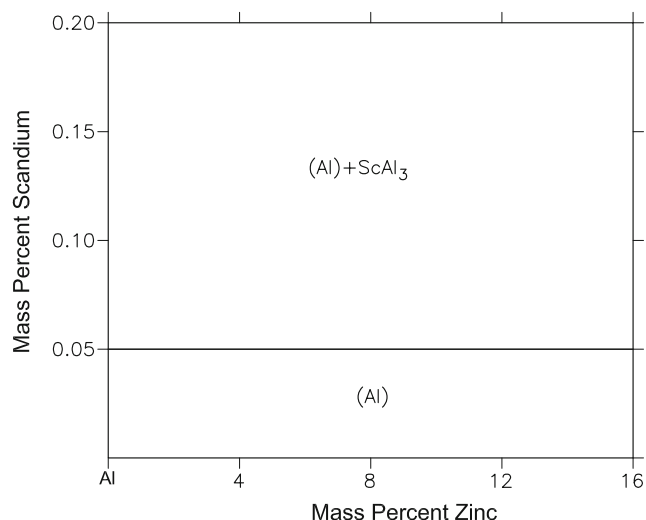


Fig. 3 Al-Sc-Zn isothermal section at 500 °C near the Al corner [2004Rok]

optical metallography, x-ray powder diffraction, electron probe microanalysis, and hardness measurements. The isothermal section constructed by [2004Rok] at 500 °C for the Al-rich region is shown in Fig. 3. The solubility of Sc in

(Al) of 0.05 mass% remains unchanged in the presence of Zn up to 15 mass%. The solubility of Zn in ScAl_3 was determined to be ~ 5 mass%. Hardness measurements by [2004Rok] showed that, similar to Al-Sc binary alloys, the ternary alloys also exhibit substantial age hardening.

References

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